# Cyclic Amidines. Part IV.* 5:6:11:12-Tetrahydro-5:11-endomethylenephenhomazine and Tröger's Base. 

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$5: 11$-Dichlorophenhomazine and its $2: 8$-dimethyl homologue are reduced
with lithium aluminium hydride to the corresponding $5: 6: 11: 12$ -
tetrahydrophenhomazines, which with formaldehyde afford $5: 6: 11: 12$ -
tetrahydro- $5: 11$-endomethylene-and $5: 6: 11: 12$-tetrahydro- $2: 8$-dimethyl-
$5:$ 11-endomethylene-phenhomazine (Tröger's base) respectively.

The ring system 5:11-endomethylene-5:6:11:12-tetrahydrophenhomazine $\{5: 11-$ $6 H: 12 H$ )-methanodibenzo $[b, f][1: 5]$ diazocine; Ring Index no. 2651\} ( $\mathrm{I} ; \mathrm{R}=\mathrm{H}$ ) has hitherto been known in the form of its 2:8-dimethyl derivative, Tröger's base ( $\mathrm{I} ; \mathrm{R}=\mathrm{Me}$ ) (Tröger, J. pr. Chem., 1887, 36, 227 ; Spielman, J. Amer. Chem. Soc., 1935, 57, 583), and a few other derivatives (Miller and Wagner, ibid., 1941, 63, 832; Smith and Schubert, ibid., 1948, 70, 2656). In confirmation of the structure proposed, Spielman (loc. cit.) prepared the base by treatment of $1: 2: 3: 4$-tetrahydro-6-methyl-3- $p$-tolylquinazoline (II; $\mathrm{R}=$ Me ) with formaldehyde. The success of this preparation appears to depend on the

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reactivity of the $2^{\prime}$-position, since the quinazolines (II; $\mathrm{R}=\mathrm{OMe}$ or OEt ) afford the corresponding analogues of Tröger's base ( $\mathrm{I} ; \mathrm{R}=\mathrm{OMe}$ or OEt ), whereas the eightmembered ring is not formed from the halogenated quinazolines (II; $\mathrm{R}=\mathrm{Cl}$ or Br ) (Miller and Wagner, loc. cit.). Wagner (J. Amer. Chem. Soc., 1935, 57, 1296) speculated on the


possibility of a more direct synthesis by the condensation of formaldehyde and $5: 6: 11: 12-$ tetrahydro-2 : 8-dimethylphenhomazine (VII; $\mathrm{R}=\mathrm{Me}$ ), but this compound was not then available. We have now investigated the synthesis of $5: 6: 11: 12$-tetrahydrophenhomazines and their conversion into their 5:11-endomethylene derivatives.


Methyl 5-methylanthranilate (III; $\mathrm{R}=\mathrm{Me}$ ) reacts exothermally with benzonitrile in the presence of " powdered " sodium to furnish (cf. Part I, J., 1954, 3429), in addition to 4-hydroxy-6-methyl-2-phenylquinazoline (IV; $\mathrm{R}=\mathrm{Me}$ ), 2 : 8-dimethyldianthranilide (V; $\mathrm{R}=\mathrm{Me}$ ) which affords its $N N^{\prime}$-dimethyl derivative (VIII) with methyl sulphate. 6:12-Dichloro-2 : 8-dimethylphenhomazine (VI; $\mathrm{R}=\mathrm{Me}$ ) results when the foregoing dianthranilide is brought into reaction with phosphorus pentachloride; experiments with dianthranilide itself ( $\mathrm{V} ; \mathrm{R}=\mathrm{H}$ ) indicated that an analogous reaction cannot be effected with thionyl chloride or phosphorus oxychloride. Interaction of the dichloro-compound (VI; $\mathrm{R}=\mathrm{Me}$ ) and sodium methoxide gives the dimethoxyphenhomazine (IX).

Schroeter (Ber., 1919, 52, 2224) has reported his inability to reduce 6:12-dichlorophenhomazine ( $\mathrm{VI} ; \mathrm{R}=\mathrm{H}$ ). We find that most of the compound is recovered unchanged after treatment of it with Raney alloy and alkali. Attempts to reduce dianthranilide with hydriodic acid without fission of the ring system were also unsuccessful. However,



dichlorophenhomazines (VI; $\mathrm{R}=\mathrm{H}$ or Me ) furnish 5:6:11:12-tetrahydrophenhomazines (VII; $\mathrm{R}=\mathrm{H}$ or Me ) when lithium aluminium hydride is employed as reducing agent. These tetrahydrophenhomazines react readily with formaldehyde to yield their 5: 11-endomethylene derivatives ( $\mathrm{I} ; \mathrm{R}=\mathrm{H}$ or Me ). The 2:8-dimethyl derivative ( $\mathrm{I} ; \mathrm{R}=\mathrm{Me}$ ) is identical with Tröger's base.

In agreement with Spielman's observations (loc. cit.), we find that, on acylation or treatment with nitrous acid, $5: 6: 11: 12$-tetrahydro-2:8-dimethyl-5:11-endomethylenephenhomazine loses the endomethylene group to afford derivatives of tetrahydrophenhomazine ( $\mathrm{X} ; \mathrm{R}=\mathrm{Me}, \mathrm{R}^{\prime}=\mathrm{Ac}, \mathrm{Bz}$, or NO ) identical with authentic specimens prepared directly from the tetrahydrophenhomazine ( $\mathrm{X} ; \mathrm{R}=\mathrm{Me}, \mathrm{R}^{\prime}=\mathrm{H}$ ). The 5 : 11-dibenzoyl ( $\mathrm{X} ; \mathrm{R}=\mathrm{H}, \mathrm{R}^{\prime}=\mathrm{Bz}$ ) and the 5 : 11-dinitroso-derivative ( $\mathrm{X} ; \mathrm{R}=\mathrm{H}, \mathrm{R}^{\prime}=\mathrm{NO}$ ) may be obtained in a similar manner both directly from the secondary amine ( $\mathrm{X} ; \mathrm{R}=\mathrm{R}^{\prime}=\mathrm{H}$ ) and, with loss of the endomethylene group, from the tertiary amine ( $\mathrm{I} ; \mathrm{R}=\mathrm{H}$ ). The
dinitrosamine ( $\mathrm{X} ; \mathrm{R}=\mathrm{H}, \mathrm{R}^{\prime}=\mathrm{NO}$ ) could not be induced to undergo the Fischer-Hepp rearrangement.

## Experimental

Spectroscopic measurements were made in absolute ethanol.
5:6:11:12-Tetrahydrophenhomazine.-There was a mildly exothermic reaction when lithium aluminium hydride ( 0.5 g .) in ether ( 80 ml .) was added to a suspension of $6: 12$-dichlorophenhomazine (Schroeter, loc. cit.) ( 2.5 g .) in ether ( 50 ml .). After 4 hours' refluxing, excess of reductant was decomposed by ethyl acetate ( 5 ml .), water ( 50 ml .) was added, and the mixture was made alkaline with sodium hydroxide ( 7 g .). The ether-soluble fraction ( 2.05 g .) afforded $5: 6: 11: 12$-tetrahydrophenhomazine ( 1.45 g ., 76\%) as colourless needles, m. p. 138.5-139.5 ${ }^{\circ}$, after two recrystallisation from light petroleum (b. p. 100-120 ) (charcoal) [Found : C, 80.2; $\mathrm{H}, \mathbf{6 . 8} ; \mathrm{N}, 13.4 \%$; $M$ (Rast), 215. $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2}$ requires C, 79.95; H, 6.7; $\mathrm{N}, 13.3 \% ; M, 210$. Light absorption : $\lambda_{\max }$ 206, 242, $290 \mathrm{~m} \mu(\varepsilon 40,000,13,400,2900)$. The dihydrochloride, obtained as small prisms by the addition of concentrated hydrochloric acid to a solution of the base in ethanol, was very sparingly soluble and could not be recrystallised; it darkened above $260^{\circ}$ and did not melt below $400^{\circ}$ (Found : C, $59.1 ; \mathrm{H}, 5 \cdot 55 ; \mathrm{N}, 9.6 . \quad \mathrm{C}_{14} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{Cl}_{2}$ requires $\mathrm{C}, 59.35 ; \mathrm{H}$, $5.7 ; \mathrm{N}, \mathbf{9 . 9} \%$ ). Interaction of the base in ethanol with picric acid gave the dipicrate as clusters of minute needles, m. p. 161-163 ${ }^{\circ}$ (Found: C, 47.0; H, 2.8; N, 16.6. $\mathrm{C}_{\mathbf{2 6}} \mathrm{H}_{\mathbf{2 0}} \mathrm{O}_{\mathbf{1 4}} \mathrm{N}_{\mathbf{8}}$ requires C, $\mathbf{4 6 . 7}$; H, $\mathbf{3 . 0}$; N, $\mathbf{1 6 . 7 5 \%}$ ).

The diacetyl derivative, prepared by boiling the base with acetic anhydride, crystallised from aqueous acetic acid as small prisms, m. p. 335- $337^{\circ}$ (Found: C, 73.5; H, 5.95; N, 9.75. $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{~N}_{2}$ requires $\mathrm{C}, \mathbf{7 3 . 4 5} ; \mathrm{H}, \mathbf{6 . 1 5} ; \mathrm{N}, \mathbf{9 . 5} \%$ ).

Benzoylation under Schotten-Baumann conditions afforded 5:11-dibenzoyl-5:6:11:12tetrahydrophenhomazine which crystallised from $n$-butanol as platelets, m. p. 306.5-307.5 ${ }^{\circ}$ (Found: C, 80.3 ; H, 5.55 ; N, 6.85. $\quad \mathrm{C}_{28} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{~N}_{2}$ requires $\mathrm{C}, 80.35 ; \mathrm{H}, 5.3 ; \mathrm{N}, 6.7 \%$ ).

5:6:11:12-Tetrahydro-5:11-dinitrosophenhomazine separated immediately on the addition of sodium nitrite to a solution of the base in dilute hydrochloric acid, and it crystallised from benzene as small prisms, m. p. 233-234 (Found: C, 62.9; H, 4.6; N, 20.2. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{O}_{2} \mathrm{~N}_{4}$ requires $\mathrm{C}, 62.65 ; \mathrm{H}, \mathbf{4 . 5}$; $\mathrm{N}, \mathbf{2 0 . 9} \%$ ). This compound gave a positive Liebermann nitrosoreaction.

5:6:11:12-Tetrahydro-5:11-endomethylenephenhomazine.-To a cold mixture of concentrated hydrochloric acid ( 5 ml .), $\mathbf{4 0} \%$ formaldehyde solution ( 2 ml .), and ethanol ( 10 ml .) was added finely powdered $5: 6: 11: 12$-tetrahydrophenhomazine ( $0 \cdot 5 \mathrm{~g}$.). The solid ( $0 \cdot 45 \mathrm{~g}$.) which was slowly deposited was collected next day, dissolved in hot water, and treated with excess of ammonia; the precipitate ( 0.35 g. ; m. p. 138-139 $)$ on crystallisation from light petroleum (b. p. 100-120 ) furnished the 5: 11-endomethylene derivative as prisms, m. p. 138$139^{\circ}$; a mixed m. p. with 5:6:11:12-tetrahydrophenhomazine was $108-115^{\circ}$. A further quantity ( $0.1 \mathrm{~g} ., \mathrm{m} . \mathrm{p} .138-139^{\circ}$ ) was recovered from the filtrate from the reaction mixture (total yield $85 \%$ ) [Found : C, 80.95 ; $\mathrm{H}, \mathbf{6 . 4}$; $\mathrm{N}, \mathbf{1 2 . 3} \%$; $M$ (Rast), 274. $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{\mathbf{2}}$ requires C, $81 \cdot 0 ; \mathrm{H}, \mathbf{6 . 3 5} ; \mathrm{N}, 12 \cdot 6 \% ; M, 222]$. Light absorption: $\lambda_{\text {max. }} 203,240,280 \mathrm{~m} \mu(\varepsilon 33,700$, 6300,1600 ). Its hydrochloride crystallised from dilute hydrochloric acid as small prisms which shrank at about $250^{\circ}$ and darkened without melting up to $360^{\circ}$; when inserted into the heating bath at $300^{\circ}$, the compound effervesced and resolidified (Found: C, 69•2; H,5.65. $\quad \mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{2} \mathrm{Cl}$ requires $\mathrm{C}, 69.6 ; \mathrm{H}, 5.85 \%$ ). The picrate was obtained as small prisms, m. p. $190-191^{\circ}$ (decomp.), from $n$-butanol (Found : C, $56.1 ; \mathrm{H}, \mathbf{3 . 6} . \mathrm{C}_{21} \mathrm{H}_{17} \mathrm{O}_{7} \mathrm{~N}_{5}$ requires C, $55.85 ; \mathrm{H}, \mathbf{3 . 8} \%$ ).

The product of treatment of the foregoing base with nitrous acid had, after crystallisation from benzene, m. p. 233-234 ${ }^{\circ}$ alone or in admixture with $5: 6: 11: 12$-tetrahydro-5:11-dinitrosophenhomazine. The compound obtained on benzoylation as described for $5: 11$-di-benzoyl-5:6:11:12-tetrahydrophenhomazine was identical with this compound as shown by $\mathrm{m} . \mathrm{p}$. and mixed m. p.

2:8-Dimethyldianthranilide.-There was an exothermic reaction when methyl 5 -methyl-
 and benzonitrile ( $\mathbf{4 8} \cdot \mathbf{4} \mathrm{g}$.) were added to "powdered " sodium ( $\mathbf{1 0 . 8} \mathrm{g}$.) in dry benzene ( $\mathbf{1 2 5} \mathrm{ml}$.). When the spontaneous boiling had subsided, the mixture was refluxed for 5 hr . Ethanol ( 10 ml .) and water ( 150 ml .) were added. The insoluble material was collected, washed with dilute hydrochloric acid and with water, and fractionally crystallised from ethanol. 2:8-Dimethyldianthranilide ( 11.6 g., $37 \%$ ) occurred as small rods, m. p. 298-299 (Found: C, 72.4; $\mathrm{H}, 5.5 ; \mathrm{N}, 10.8 . \mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}_{2} \mathrm{~N}_{2}$ requires $\mathrm{C}, \mathbf{7 2 . 1 5} ; \mathrm{H}, 5 \cdot 3 ; \mathrm{N}, 10.5 \%$ ). Light absorption : $\lambda_{\text {max. }} 210 \mathrm{~m} \mu(\varepsilon 35,600)$. This compound was but sparingly soluble in aqueous sodium hydroxide.

The second product, 4-hydroxy-6-methyl-2-phenylquinazoline ( $5 \cdot 7 \mathrm{~g}$.), separated as pale yellow needles, m. p. 265-266 ${ }^{\circ}$ (Found: C, $76.5 ; \mathrm{H}, 4.9 ; \mathrm{N}, 12.1 . \mathrm{C}_{15} \mathrm{H}_{12} \mathrm{ON}_{2}$ requires C, 76.25 ; H, $5 \cdot 1 ; \mathrm{N}, 11 \cdot 85 \%)$. Light absorption : $\lambda_{\text {max. }} 209,237,294 \mathrm{~m} \mu(\varepsilon 28,700,27,700,16,000)$. Light absorption of 4-hydroxy-2-phenylquinazoline : $\lambda_{\max .} 206,237,291 \mathrm{~m} \mu(\varepsilon 25,100,26,500,14,300)$. A further quantity ( $3 \cdot 1 \mathrm{~g}$.) of the quinazoline was recovered by faintly acidifying the aqueous layer obtained in working up.
$2: 8: \mathrm{N}: \mathrm{N}^{\prime}$-Tetramethyldianthranilide, prepared by treatment of the foregoing dianthranilide ( 0.4 g .) with methyl sulphate ( 0.5 ml .) and 2 N -sodium hydroxide ( 10 ml .), crystallised from benzene-light petroleum (b. p. $80-100^{\circ}$ ) as needles, m. p. 254-255 (Found: C, 73.7; H, 6.1. $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{~N}_{2}$ requires $\mathrm{C}, 73 \cdot 45 ; \mathrm{H}, 6 \cdot 15 \%$ ). Light absorption : $\lambda_{\text {max }} .213 \mathrm{~m} \mathrm{\mu}(\varepsilon 31,100)$

6:12-Dichloro-2:8-dimethylphenhomazine.-Finely powdered 2:8-dimethyldianthranilide ( 7.45 g .) and phosphorus pentachloride ( 12.5 g .) were boiled together in chloroform ( 50 ml .) for 4 hr . After filtration, the solid ( 4.85 g .) which separated on concentration of the filtrate was crystallised from light petroleum (b. p. $100-120^{\circ}$ ) and afforded $6: 12$-dichloro-2 : 8-dimethylphenhomazine ( $4.05 \mathrm{~g} ., 48 \%$ ) as prisms, m. p. 209-210 ${ }^{\circ}$ (Found: C, 63.7; H, 3.65; N, 9.3. $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{Cl}_{2}$ requires $\mathrm{C}, 63 \cdot 35 ; \mathrm{H}, 4 \cdot 0 ; \mathrm{N}, 9 \cdot 25 \%$ ).

6:12-Dimethoxy-2:8-dimethylphenhomazine was formed in $90 \%$ yield by boiling the foregoing dichloro-compound in methanolic sodium methoxide for 23 hr ., and it crystallised from light petroleum (b. p. 100- $120^{\circ}$ ) as prisms, m. p. $155-156^{\circ}$ (Found : C, 73.4 ; H, 5.9; N, 9.4 . $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{~N}_{2}$ requires C, 73.45 ; $\mathrm{H}, 6.15 ; \mathrm{N}, 9.5 \%$. Light absorption: $\lambda_{\text {max. }} 204,258 \mathrm{~m} \mu$ $(\varepsilon 47,100,6600)$. Light absorption of 6:12-dimethoxyphenhomazine : $\lambda_{\max }$. $203,253 \mathrm{~m} \mu$ $(\varepsilon 46,100,5900)$ and of $6: 12$-diethoxyphenhomazine : $\lambda_{\max } 203,252 \mathrm{~m} \mu(\varepsilon 45,500,6100)$; the values for these two compounds recorded in Part I (loc. cit.) are incorrect.
$5: 6: 11: 12-T e t r a h y d r o-2: 8$-dimethylphenhomazine.-After $6: 12$-dichloro-2:8-dimethylphenhomazine ( 2.7 g .) and lithium aluminium hydride ( 0.5 g .) had been brought into reaction for $2 \nmid \mathrm{hr}$. as described for $5: 6: 11: 12$-tetrahydrophenhomazine, the ether-soluble material ( 1.85 g .) was separated by treatment with aqueous lactic acid into a neutral fraction ( 0.6 g .) , which was unchanged starting material, and a basic fraction ( $0.8 \mathrm{~g} ., \mathrm{m} . \mathrm{p} .197-201^{\circ}$ ). The latter on crystallisation from light petroleum (b. p 100-120 ) afforded 5:6:11:12-tetrahydro2: 8-dimethylphenhomazine as colourless leaflets, m. p. 205-206 ${ }^{\circ}$, depressed to below $190^{\circ}$ by the original dichloro-compound (Found: $\mathrm{C}, 80.6 ; \mathrm{H}, 7.65 ; \mathrm{N}, 11.7 . \mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2}$ requires C , $80 \cdot 6 ; \mathrm{H}, 7.6 ; \mathrm{N}, 11.75 \%$ ). Light absorption : $\lambda_{\text {max. }} 207,242,295 \mathrm{~m} \mu(\varepsilon 46,900,16,200,3200)$. Prolongation of the period of reduction to $8 \frac{1}{2} \mathrm{hr}$. gave the desired product in $74 \%$ yield. Its dipicrate separated as small prisms, m. p. 162-164 ${ }^{\circ}$, when the base and picric acid reacted in ethanol (Found: C, $48 \cdot 1 ; \mathrm{H}, 3 \cdot 4 . \quad \mathrm{C}_{28} \mathrm{H}_{24} \mathrm{O}_{14} \mathrm{~N}_{8}$ requires $\mathrm{C}, 48 \cdot 3 ; \mathrm{H}, \mathbf{3 . 4 5} \%$ ).

Acetylation of the base with boiling acetic anhydride afforded 5:11-diacetyl-5:6:11:12-tetrahydro-2:8-dimethylphenhomazine which separated as small prisms, m. p. 297-298 ${ }^{\circ}$, from aqueous acetic acid (Found : N, 8.85. Calc. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{~N}_{2}: \mathrm{N}, 8.7 \%$ ). This compound did not depress the m . p. of the product of acetylation of Tröger's base, m. p. 297-298 ${ }^{\circ}$, obtained by the method of Spielman who records (loc. cit.) m. p. 286-288 ${ }^{\circ}$.

The dibenzoyl derivative, prepared under Schotten-Baumann conditions, occurred as platelets, m. p. $301-302^{\circ}$, from xylene (Found : C, $80.6 ; \mathrm{H}, 5 \cdot 9$. Calc. for $\mathrm{C}_{30} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{~N}_{2}$ : C, 80.7 ; H, $5.85 \%$ ). By this method, the same compound, m. p. and mixed m. p. $301-302^{\circ}$, was obtained from Tröger's base in $56 \%$ yield; by Spielman's method the yield was only $10 \%$. Spielman records m. p. 290-291 for this compound.

On being treated with nitrous acid, this base afforded the $5: 11$-dinitroso-derivative, $\mathrm{m} . \mathrm{p}$. $247-248^{\circ}$, undepressed on admixture with the compound obtained by Spielman's method from Tröger's base (Found : C, 65.0; H, 5.7. Calc. for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{O}_{2} \mathrm{~N}_{4}$ : C, 64.85; H, 5.45\%). Spielman records m. p. 246-247 ${ }^{\circ}\left(254-255^{\circ}\right.$ corr.) for this compound.
$5: 6: 11: 12-T e t r a h y d r o-2: 8$-dimethyl-5:11-endomethylenephenhomazine, prepared from the corresponding tetrahydrophenhomazine and formaldehyde in $95 \%$ yield, was obtained as colourless needles, m. p. 136-137 ${ }^{\circ}$, undepressed on admixture with Tröger's base prepared in $30 \%$ yield by Goecke's method (Z. Elektrochem., 1903, 9, 470). Light absorption : $\lambda_{\max }$ 205, 235, $285 \mathrm{~m} \mu(\varepsilon 35,700,8300,2000)$; in hexane, $\lambda_{\max } .246,290 \mathrm{~m} \mu(\varepsilon 8200,2200)$. Wepster (Rec. Trav. chim., 1953, 72, 661) records for Tröger's base in isooctane: $\lambda_{\max } 247.5$, $291 \mathrm{~m} \mu$ ( $\varepsilon 8490$, 2230).

